In the Claims

Please amend the claims as follows:

- **6.** (original) A solid single-phase amorphous stabiliser, which comprises 2 or more compounds having a molecular weight in the range from 300-1000 g/mol.
- 7. (original) An amorphous solid or subcooled melt of

2,2'-methylenebis(4-[1,1,3,3-tetramethylbutyl]-6-benzotriazol-2-yl-phenol);

bis(2-methyl-4-hydroxy-5-tert-butylphenyl)sulfide;

N,N'-bis(3-[3',5'-di-tert-butyl-4'-hydroxyphenyl]propionyl)hexamethylenediamine;

1,3,5-trimethyl-2,4,6-tris(3',5'-di-tert-butyl-4'-hydroxybenzyl)benzene;

1,3,5-tris(3,5-di-tert-butyl-4-hydroxybenzyl)-1,3,5-triazine-2,4,6(1H,3H,5H)-trione;

 $1,3,5-tris(4-tert-butyl-3-hydroxy-2,6-dimethylbenzyl)\\ 1,3,5-triazine-2,4,6-(1H,3H,5H)-trione;$

di(1,2,2,6,6-pentamethylpiperidin-4-yl)-2-(3,5-di-tert-butyl-4-hydroxybenzyl)-2-n-butylmalonate;

2-(2'-hydroxy-3',5'-bis(1,1-dimethylbenzyl)phenyl)benzotriazole;

2-(2'-hydroxy-3',5'-di-tert-butylphenyl)benzotriazole;

2-(2-hydroxy-3,5-di-tert-butylphenyl)-5-chlorobenzotriazole;

an isomeric mixture of 5,7-di-tert-butyl-3-(3,4-dimethylphenyl)-(9d)-2(3H)-benzofuranone and 5,7-di-

tert-butyl-3-(2,3-dimethylphenyl)-(9d)-2(3H)-benzofuranone;

or of a compound of formula

$$(CH_3)_3C$$

$$HO$$

$$C(CH_3)_3$$

$$HO$$

$$C(CH_3)_3$$

$$C(CH_3)_3$$

- **8.** (original) A mixture comprising a subcooled melt or amorphous solid of a compound according to claim **7** in an amount from 5 to 100 % by weight.
- **9.** (previously amended) A process for the preparation of a subcooled melt according to claim **7**, which comprises rapidly cooling the melt to a temperature from the regular melting point to the glass transition temperature of the homogeneous phase.
- **10.** (currently amended) A process for the preparation of <u>anthe</u> amorphous solid according to claim **7**, which comprises chilling the melt or subcooled melt-according to claim **7** to a temperature below its glass transition temperature.
- 11. (original) Crystalline B-modification of the compound 14

$$CI$$
 N
 OH
 H_3C
 CH_3
 H_3C
 CH_3
 CH_3

characterised by the interplanar spacings are $9.4\cdot10^{-10}\,\mathrm{m}$, $4.69\cdot10^{-10}\,\mathrm{m}$, $3.94\cdot10^{-10}\,\mathrm{m}$ and $3.79\cdot10^{-10}\,\mathrm{m}$.

12. (currently amended) A mixture consisting of different modifications of compound 14 of formula

which comprises from 40 up to 100 % by weight of B-crystalline form according to claim 11.

- 13. (original) A process for the preparation of the ß-crystalline form of compound 14 according to claim 11, which comprises crystallising, recrystallising or tempering compound 14 in the temperature range from 95°C to the melting point and then cooling it rapidly.
- **14. (original)** A process for colour-stabilising amorphous pentaerythritol-tetrakis(3-[3',5'-di-tert-butyl-4'-hydroxyphenyl]propionate), which comprises admixing a stabiliser from the class consisting of the organic phosphites, phosphonites and/or benzofuran-2-ones to a melt consisting of pentaerythritol-tetrakis(3-[3',5'-di-tert-butyl-4'-hydroxyphenyl]propionate) and solidifying the mixture so obtained.
- **15. (original)** Colour-stabilised amorphous pentaerythritol-tetrakis(3-[3',5'-di-tert-butyl-4'-hydroxy-phenyl]propionate) containing a sufficient stabilising amount of a stabiliser from the class consisting of the organic phosphites, phosphonites and/or benzofuran-2-ones.
- 16. (original) A stabiliser composition, which comprises
 - a) pentaerythritol-tetrakis(3-[3',5'-di-tert-butyl-4'-hydroxyphenyl]propionate), and
 - b) a compound of the benzofuran-2-one type.

- 17. (previously amended) A composition, which comprises
- A) an organic material susceptible to oxidative, thermal or/and actinic degradation or build-up, and
- B) the amorphous form of a compound according to claim 7, the β-crystalline form of compound 14

$$CI$$
 N
 OH
 CH_3
 CH_3
 H_3C
 CH_3
 CH_3

characterised by interplanar spacings at 9.4·10⁻¹⁰ m, 4.69·10⁻¹⁰ m, 3.94·10⁻¹⁰ m and 3.79·10⁻¹⁰ m and/or the stabiliser composition which comprises a) pentaerythritol-tetrakis(3-[3',5'-di-tert-butyl-4'-hydroxyphenyl]propionate), and b) a compound of the benzofuran-2-one type as stabiliser.

18. (previously amended) A process for stabilising organic material against oxidative, thermal or actinic degradation or build-up, which comprises adding to the material the amorphous form of a compound according to claim **7**, the β-crystalline form of compound 14

$$CI$$
 N
 N
 H_3C
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3

characterised by interplanar spacings at 9.4 10⁻¹⁰ m, 4.69 10⁻¹⁰ m, 3.94 10⁻¹⁰ m and 3.79 10⁻¹⁰ m and/or the stabiliser composition which comprises a) pentaerythritol-tetrakis(3-[3',5'-di-tert-butyl-4'-hydroxyphenyl]propionate), and b) a compound of the benzofuran-2-one type as stabiliser.

- **19. (original)** A process according to claim **18**, wherein 0.01 to 15 parts by weight of the stabiliser are added to 100 parts by weight of organic material to be stabilised.
- **20.** (original) A process according to claim 18, wherein the organic material is a synthetic thermoplastic polymer.
- **21.** (original) A process according to claim 18, wherein a customary additive is added as further component.
- **22.** (**previously added**) Granules obtained by a process which comprises extruding a subcooled melt consisting essentially of an organic compound having a molecular weight of 200 to 1500 g/mol, or the plastic composition consisting of the mixture of the subcooled melt and a further component, which is selected from compounds of the subcooled melt in crystalline form and other conventional additives.
- **23.** (currently amended) A process for the preparation of a subcooled melt of a mixture comprising a subcooled melt according to claim **8**, which comprises rapidly cooling the melt to a temperature from the regular melting point to the glass transition temperature of the homogeneous phase.
- **24.** (currently amended) A process for the preparation of a mixture comprising an amorphous solid according to claim **8**, which comprises chilling the melt or a subcooled melt according to claim **8** to a temperature below its glass transition temperature.